

# CARBON DIOXIDE, REFRIGERATED

Safety Data Sheet

# 1. IDENTIFICATION

Product identifier

Product Name CARBON DIOXIDE, REFRIGERATED LIQUID

Other means of identification

Safety data sheet number IOC-P024 UN/ID no. UN2187

Synonyms Carbonic Anhydride, Refrigerated Liquid

Recommended use of the chemical and restrictions on use

Recommended Use Industrial and professional use.

Uses advised against Consumer use

# Details of the supplier of the safety data sheet

Indiana Oxygen Company 6099 W. Corporate Way Indianapolis, IN 46278 Phone: 317-290-0003 wwwe.indianaoxygen.com

For additional product information contact your local customer service.

Emergency telephone number

Company Phone Number 1-800-535-5053 (Infotrak)

<sup>\*</sup> May include subsidiaries or affiliate companies/divisions.

# 2. HAZARDS IDENTIFICATION

## Classification

## **OSHA Regulatory Status**

This chemical is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200).

Gases under pressure	Refrigerated liquefied gas
Simple asphyxiants	Yes

## Label elements



Signal word

Warning

Hazard Statements Contains refrigerated gas; may cause cryogenic burns or injury May displace oxygen and cause rapid suffocation

May increase respiration and heart rate

Precautionary Statements - Prevention
Do not handle until all safety precautions have been read and understood
Use and store only outdoors or in a well ventilated place
Wear cold insulating gloves, face shield, and eye protection
Use a backflow preventive device in piping
Do NOT change or force fit connections
Close valve after each use and when empty
Always keep container in upright position

Precautionary Statements - Response

IF INHALED: Remove person to fresh air and keep comfortable for breathing. Get medical attention/advice.

IF ON SKIN:. Thaw frosted parts with lukewarm water. Do not rub affected area. Get immediate medical advice/attention.

Hazards not otherwise classified (HNOC)

Not applicable

# 3. COMPOSITION/INFORMATION ON INGREDIENTS

Chemical Name	CAS No.	Volume %	Chemical Formula
Carbon dioxide	124-38-9	100	CO 2

# 4. FIRST AID MEASURES

## Description of first aid measures

General advice Show this safety data sheet to the doctor in attendance.

Inhalation Remove to fresh air and keep comfortable for breathing. If breathing is difficult, give oxygen. If

breathing has stopped, give artificial respiration. Get medical attention immediately.

Skin contact For dermal contact or suspected frostbite, remove contaminated clothing and flush affected areas

with lukewarm water. DO NOT USE HOT WATER. A physican should see the patient promptly if contact with the product has resulted in blistering of the dermal surface or in deep tissue freezing.

Eye contact If frostbite is suspected, flush eyes with cool water for 15 minutes and obtain immediate medical

attention.

Ingestion Not an expected route of exposure.

Self-protection of the first aider RESCUE PERSONNEL SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS.

## Most important symptoms and effects, both acute and delayed

Symptoms Simple asphyxiant. May cause suffocation by displacing the oxygen in the air. Exposure to

oxygen-deficient atmosphere (<19.5%) may cause dizziness, drowsiness, nausea, vomiting, excess salivation, diminished mental alertness, loss of consciousness and death. Exposure to atmospheres containing 8-10% or less oxygen will bring about unconsciousness without warning and so quickly that the individuals cannot help or protect themselves. Lack of sufficient oxygen may cause serious injury or death. Depending on concentration and duration of exposure to carbon dioxide may cause increased respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric oxygen is

decreased to 15-17%. Contact with liquid may cause cold burns/frostbite.

## Indication of any immediate medical attention and special treatment needed

## 5. FIRE-FIGHTING MEASURES

#### Suitable extinguishing media

Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

#### Specific extinguishing methods

Continue to cool fire exposed cylinders until flames are extinguished. Damaged cylinders should be handled only by specialists.

## Specific hazards arising from the chemical

Non-flammable gas. Cylinders may rupture under extreme heat.

## <u>Protective equipment and precautions for firefighters</u>

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear.

## 6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Personal precautions Evacuate personnel to safe areas. Ensure adequate ventilation, especially in confined areas.

Monitor oxygen level. Wear self-contained breathing apparatus when entering area unless atmosphere is proved to be safe. Use personal protection recommended in Section 8.

Other Information When in contact with refrigerated/cryogenic liquids, many materials become brittle and are likely

to break without warning.

**Environmental precautions** 

Environmental precautions Prevent spreading of vapors through sewers, ventilation systems and confined areas.

Methods and material for containment and cleaning up

Methods for containment Stop the flow of gas or remove cylinder to outdoor location if this can be done without risk. If leak is

in container or container valve, contact the appropriate emergency telephone number in Section 1 or call your closest

Indiana Oxygen Company location.

Methods for cleaning up Return Portable Cryogenic Container to Indiana Oxygen or an authorized distributor.

## 7. HANDLING AND STORAGE

## Precautions for safe handling

Advice on safe handling

Never allow any unprotected part of the body to touch uninsulated pipes or vessels that contain cold fluids. The extremely cold metal will cause moist flesh to stick fast and tear when one attempts to withdraw from it. Do NOT change or force fit connections. For applications with moist Carbon Dioxide, 316, 309 and 310 stainless steels may be used as well as Hastelloy® A, B, & C and Monel®. Ferrous nickel alloys are slightly suspectible to corrosion. At normal temperatures carbon dioxide is compatible with most plastics and elastomers.

Protect cylinders from physical damage; do not drag, roll, slide or drop. When moving cylinders, even for short distance, use a cart designed to transport cylinders. Never attempt to lift a cylinder by its valve protection cap. Never insert an object (e.g. wrench, screwdriver, pry bar,etc.) into valve cap openings. Doing so may damage valve, causing leak to occur. Use an adjustable strap wrench to remove over-tight or rusted caps. If user experiences any difficulty operating cylinder valve discontinue use and contact supplier. Use only with adequate ventilation. Use a backflow preventive device in piping. Close valve after each use and when empty. Ensure the complete gas system has been checked for leaks before use.

Never put cylinders into trunks of cars or unventilated areas of passenger vehicles. Never strike an arc on a compressed gas cylinder or make a cylinder a part of an electrical circuit. Never attempt to refill a compressed gas cylinder without the owner's written consent.

Only experienced and properly instructed persons should handle gases under pressure. Always store and handle compressed gas cylinders in accordance with Compressed Gas Association, pamphlet CGA-P1, Safe Handling of Compressed Gases in Containers. Use only with equipment rated for cylinder pressure.

For additional storage recommendations, consult Compressed Gas Association's Pamphlets P-1,AV-7, G-6, G-6.1, G-6.2, G6.3, G-6.5, G-6.7, G-6.9, PS-5,TB-10, and SB-2.

# Conditions for safe storage, including any incompatibilities

Storage Conditions Store in coo

Store in cool, dry, well-ventilated area of non-combustible construction away from heavily trafficked areas and emergency exits. Cylinders should be stored upright with valve protection cap in place and firmly secured to prevent falling. Keep at temperatures below 52°C / 125°F. Full and empty cylinders should be segregrated. Use a "first in-first out" inventory system to prevent full cylinders from being stored for excessive periods of time. Stored containers should be periodically checked for general condition and leakage.

Incompatible materials

Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.

## 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

## Control parameters

Exposure Guidelines

Chemical Name	ACGIH TLV	OSHA PEL	NIOSH IDLH
Carbon dioxide	STEL = 30000 ppm	TWA: 5000 ppm	IDLH: 40000 ppm
124-38-9	TWA: 5000 ppm	TWA: 9000 mg/m <sup>3</sup>	TWA: 5000 ppm
		(vacated) TWA: 10000 ppm	TWA: 9000 mg/m <sup>3</sup>
		(vacated) TWA: 18000 mg/m <sup>3</sup>	STEL: 30000 ppm
		(vacated) STEL: 30000 ppm	STEL: 54000 mg/m <sup>3</sup>
		(vacated) STEL: 54000 mg/m³	

ACGIH TLV: American Conference of Governmental Industrial Hygienists - Threshold Limit Value. OSHA PEL: Occupational Safety and Health Administration - Permissible Exposure Limits. NIOSH IDLH: Immediately Dangerous to Life or Health Immediately Dangerous to Life or Health.

Other Information Vacated limits revoked by the Court of Appeals decision in AFL-CIO v. OSHA, 965 F.2d 962 (11th Cir.,

1992).

Appropriate engineering controls

Engineering Controls Local exhaust ventilation to prevent accumulation of high concentrations and maintain air-oxygen

levels at or above 19.5%. Oxygen detectors should be used when asphyxiating gases may be

released.

Individual protection measures, such as personal protective equipment

Eye/face protection Wear safety glasses with side shields (or goggles). If splashes are likely to occur, wear:. Goggles.

Face-shield.

Skin and body protection Work gloves and safety shoes are recommended when handling cylinders. Wear cold insulating

gloves when handling liquid.

Respiratory protection Use positive pressure airline respirator with escape cylinder or self contained breathing apparatus

for oxygen-deficient atmospheres (<19.5%). If exposure limits are exceeded or irritation is experienced, NIOSH/MSHA approved respiratory protection should be worn. Positive-pressure supplied air respirators may be required for high airborne contaminant concentrations. Respiratory

protection must be provided in accordance with current local regulations.

General Hygiene Considerations Handle in accordance with good industrial hygiene and safety practice. Do not get in eyes, on skin,

or on clothing.

## 9. PHYSICAL AND CHEMICAL PROPERTIES

#### Information on basic physical and chemical properties

Physical state Refrigerated liquefied gas

Appearance Colorless.
Odor Odorless.

Odor threshold No information available

pHNo data availableMelting pointNo data availableEvaporation rateNot applicable

Lower flammability limit:Not applicableUpper flammability limit:Not applicableFlash pointNot applicableAutoignition temperatureNo data availableDecomposition temperatureNo data available

Water solubility
Partition coefficient
Kinematic viscosity
No data available
No data available
No data available
No data available
Not applicable

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Chemical Name	Molecular weight	Boiling point	Vapor Pressure	Vapor density (air =1)	Gas Density kg/m³@20°C	Critical Temperature
Carbon dioxide	44.01	-78.5 °C (Sublimes)	838 psig (5778 kPa) @ 21.1°C	1.522	1.839	31.1 °C

## 10. STABILITY AND REACTIVITY

#### Reactivity

Not reactive under normal conditions

#### Chemical stability

Stable under normal conditions.

## Explosion data

Sensitivity to Mechanical Impact None.
Sensitivity to Static Discharge None.

## Possibility of Hazardous Reactions

None under normal processing.

## Conditions to avoid

Due to the presence of Carbon dioxide, Carbonic acid is formed in the presence of moisture.

## **Incompatible materials**

Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.

#### **Hazardous Decomposition Products**

Oxygen. Carbon monoxide (CO).

## 11. TOXICOLOGICAL INFORMATION

## Information on likely routes of exposure

Inhalation Acidosis, adrenal cortical exhaustion, and other metabolic stresses have resulted from prolonged

continuous exposure to 1-2% carbon dioxide (10,000 ppm-20,000 ppm). The ACGIH TLV of 5,000 ppm is expected to provide a good margin of safety from asphyxiation and undue metabolic stress provided sufficient oxygen levels are maintained in the air. Increased physical activity, duration of exposure, and decreased oxygen content can affect systemic and respiratory effects resulting from

exposure to carbon dioxide.

Skin contact Contact with liquid may cause cold burns/frostbite.

Eye contact Contact with liquid may cause cold burns/frostbite.

Ingestion Not an expected route of exposure.

Information on toxicological effects

Symptoms Depending on concentration and duration of exposure to carbon dioxide may cause increased

respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric oxygen is

decreased to 15-17%.

Delayed and immediate effects as well as chronic effects from short and long-term exposure

IrritationNot classified.SensitizationNot classified.Germ cell mutagenicityNot classified.

Carcinogenicity This product does not contain any carcinogens or potential carcinogens listed by OSHA, IARC or NTP.

Reproductive toxicity
STOT - single exposure
STOT - repeated exposure
Not classified.
Not classified.

Chronic toxicity Chronic harmful effects are not known from repeated inhalation of concentrations below PEL/TLV.

Target Organ Effects Central Vascular System (CVS), Respiratory system.

Aspiration hazard Not applicable.

#### Numerical measures of toxicity

Chemical Name	Oral LD50	Dermal LD50	Inhalation LC50	Inhalation LC50 (CGA P-20)
Carbon dioxide	-	-	470,000 ppm (Rat)	-
124-38-9				

Product Information

Oral LD50

Dermal LD50 No information available.

Inhalation LC50 TCLo - 10,000 ppm (Rat) 24 hours/30 days-continuous

Inhalation LC50 No information available.

# 12. ECOLOGICAL INFORMATION

**Ecotoxicity** 

No known acute aquatic toxicity.

Persistence and degradability

No information available.

Bioaccumulation

No information available.

Other adverse effects

Can cause frost damage to vegetation.

Global warming potential (GWP) 1

# 13. DISPOSAL CONSIDERATIONS

Waste treatment methods

Disposal of wastes

Do not attempt to dispose of residual waste or unused quantities. Return in the shipping container PROPERLY

LABELED WITH ANY VALVE OUTLET PLUGS OR CAPS SECURED AND VALVE PROTECTION CAP IN PLACE to Indiana Oxygen

for proper disposal.

# 14. TRANSPORT INFORMATION

DOT

UN/ID no. UN2187

Proper shipping name Carbon dioxide, refrigerated liquid

Hazard Class 2.2 Special Provisions T75, TP5

Description UN2187, Carbon dioxide, refrigerated liquid, 2.2

Emergency Response Guide Number 120

TDG

UN/ID no. UN2187

Proper shipping name Carbon dioxide, refrigerated liquid

Hazard Class 2.2

Description UN2187, Carbon dioxide, refrigerated liquid, 2.2

MEX

UN/ID no. UN2187

Proper shipping name Carbon dioxide, refrigerated liquid

Hazard Class 2.3

Description UN2187, Carbon dioxide, refrigerated liquid, 2.3

IATA

UN/ID no. UN2187

Proper shipping name Carbon dioxide, refrigerated liquid

Hazard Class 2.2 ERG Code 2L

Description UN2187, Carbon dioxide, refrigerated liquid, 2.2

**IMDG** 

UN/ID no. UN2187

Proper shipping name Carbon dioxide, refrigerated liquid

Hazard Class 2.2 EmS-No. F-C, S-V

Description UN2187, Carbon dioxide, refrigerated liquid, 2.2

**ADR** 

UN/ID no. UN2187

Proper shipping name Carbon dioxide, refrigerated liquid

Hazard Class 2.2
Classification code 3A
Tunnel restriction code (C/E)
Special Provisions 593

Description UN2187, Carbon dioxide, refrigerated liquid, 2.2, (C/E)

## 15. REGULATORY INFORMATION

International Inventories

TSCA Complies
DSL/NDSL Complies
EINECS/ELINCS Complies

Legend:

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory

DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List

EINECS/ELINCS - European Inventory of Existing Chemical Substances/European List of Notified Chemical Substances

## **US Federal Regulations**

#### **SARA 313**

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product does not contain any chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372.

SARA 311/312 Hazard Categories

Acute Health Hazard Yes
Chronic Health Hazard No
Fire Hazard No
Sudden release of pressure hazard Yes
Reactive Hazard No

## **CERCLA**

This material, as supplied, does not contain any substances regulated as hazardous substances under the Comprehensive Environmental Response Compensation and Liability Act (CERCLA) (40 CFR 302) or the Superfund Amendments and Reauthorization Act (SARA) (40 CFR 355). There may be specific reporting requirements at the local, regional, or state level pertaining to releases of this material.

#### Clean Air Act, Section 112 Hazardous Air Pollutants (HAPs) (see 40 CFR 61)

This product does not contain any substances regulated as hazardous air pollutants (HAPS) under Section 112 of the Clean Air Act Amendments of 1990.

## CWA (Clean Water Act)

This product does not contain any substances regulated as pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42)

## Risk and Process Safety Management Programs

This material, as supplied, does not contain any regulated substances with specified thresholds under 40 CFR Part 68. This product does not contain any substances regulated as Highly Hazardous Chemicals pursuant to the 29 CFR Part 1910.110.

# **US State Regulations**

## California Proposition 65

This product does not contain any Proposition 65 chemicals

## U.S. State Right-to-Know Regulations

Chemical Name	New Jersey	Massachusetts	Pennsylvania
Carbon dioxide	Х	X	Х
124-38-9			

## **International Regulations**

Chemical Name	Carcinogenicity	Exposure Limits	
Carbon dioxide	-	Mexico: TWA= 5000 ppm	
		Mexico: TWA= 9000 mg/m <sup>3</sup>	
		Mexico: STEL= 15000 ppm	
		Mexico: STEL= 27000 mg/m <sup>3</sup>	

## **16. OTHER INFORMATION**

NFPA Health hazards 3 Flammability 0 Instability 0 Physical and Chemical

Properties Simple

asphyxiant

Note: Ratings were assigned in accordance with Compressed Gas Association (CGA) guidelines as published in CGA Pamphlet P-19-2009, CGA Recommended Hazard Ratings for Compressed Gases, 3rd Edition.

Issue Date17-Feb-2015Revision Date19-May-2015Revision NoteInitial Release.

#### **General Disclaimer**

For terms and conditions, including limitation of liability, please refer to the purchase agreement in effect between Indiana Oxygen Company (or any of their affiliates and subsidiaries) and the purchaser.

## DISCLAIMER OF EXPRESSED AND IMPLIED WARRANTIES

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**End of Safety Data Sheet**