Revision Date 24-Jun-2015



# CARBON DIOXIDE (2-25%) In ARGON Safety Data Sheet

# **1. IDENTIFICATION**

Product identifier Product Name

CARBON DIOXIDE (2-25%) In ARGON

Other means of identification Safety data sheet number UN/ID no. Trade name

IOC-M0027 UN1956 CORGON 5,CORGON 8, CORGON 10, CORGON 15, CORGON 18, CORGON 20, CORGON 25, CRONIGON 2.5, ARGOSHIELD 8C; ARGOSHIELD 25C; ARGOSHIELD GP

Recommended use of the chemical and restrictions on useRecommended UseIndustrial and professional use.Uses advised againstConsumer use

# Details of the supplier of the safety data sheet

Indiana Oxygen Company 6099 Corporate Way Indianapolis, IN 46278 Phone: (317) 290-0003 www.Indianaoxygen.com

\* May include subsidiaries or affiliate companies/divisions.

For additional product information contact your local customer service.

Emergency telephone number

Company Phone Number 1-800-535-5053 (Infotrak)

# 2. HAZARDS IDENTIFICATION

# **Classification**

# **OSHA Regulatory Status**

This chemical is considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200).

Gases under pressure	Compressed gas
	Yes

#### Label elements



Signal word

Warning

Hazard Statements Contains gas under pressure; may explode if heated May displace oxygen and cause rapid suffocation

May increase respiration and heart rate

Precautionary Statements - Prevention Do not handle until all safety precautions have been read and understood Avoid breathing gas Use and store only outdoors or in a well ventilated place Use a backflow preventive device in piping Use only with equipment rated for cylinder pressure Close valve after each use and when empty

Precautionary Statements - Response IF INHALED: Remove person to fresh air and keep comfortable for breathing. Get medical attention/advice.

Precautionary Statements - Storage Protect from sunlight when ambient temperature exceeds 52°C/125°F

Hazards not otherwise classified (HNOC) Not applicable

# **3. COMPOSITION/INFORMATION ON INGREDIENTS**

Chemical Name	CAS No.	Volume %	Chemical Formula
Argon	7440-37-1	60 - 100	Ar
Carbon dioxide	124-38-9	2-25	CO 2

Composition listed covers broad ranges rather than exact percentages for specific products.

# 4. FIRST AID MEASURES

Description of first aid measures

General advice	Show this safety data sheet to the doctor in attendance.
Inhalation	Remove to fresh air and keep comfortable for breathing. If breathing is difficult, give oxygen. If breathing has stopped, give artificial respiration. Get medical attention immediately.
Skin contact	None under normal use. Get medical attention if symptoms occur.
Eye contact	None under normal use. Get medical attention if symptoms occur.
Ingestion	Not an expected route of exposure.
Self-protection of the first aider	RESCUE PERSONNEL SHOULD BE EQUIPPED WITH SELF-CONTAINED BREATHING APPARATUS.
Most important symptoms and effects, b	both acute and delayed
Symptoms	Simple asphyxiant. May cause suffocation by displacing the oxygen in the air. Exposure to oxygen-deficient atmosphere (<19.5%) may cause dizziness, drowsiness, nausea, vomiting, excess salivation, diminished mental alertness, loss of consciousness and death. Exposure to atmospheres containing 8-10% or less oxygen will bring about unconsciousness without warning and so quickly

that the individuals cannot help or protect themselves. Lack of sufficient oxygen may cause serious injury or death. Depending on concentration and duration of exposure to carbon dioxide may cause increased respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric oxygen is decreased to 15-17%.

Indication of any immediate medical attention and special treatment needed

Note to physicians

Treat symptomatically.

# **5. FIRE-FIGHTING MEASURES**

Suitable extinguishing media

Use extinguishing measures that are appropriate to local circumstances and the surrounding environment.

Specific extinguishing methods

Continue to cool fire exposed cylinders until flames are extinguished. Damaged cylinders should be handled only by specialists.

Specific hazards arising from the chemical

Non-flammable gas. Cylinders may rupture under extreme heat.

Protective equipment and precautions for firefighters

As in any fire, wear self-contained breathing apparatus pressure-demand, MSHA/NIOSH (approved or equivalent) and full protective gear.

# 6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures

Personal precautions Evacuate personnel to safe areas. Ensure adequate ventilation, especially in confined areas. Monitor oxygen level. Wear self-contained breathing apparatus when entering area unless atmosphere is proved to be safe.

Environmental precautions

Environmental precautions	Prevent spreading of vapors through sewers, ventilation systems and confined areas.
Methods and material for containr	nent and cleaning up
Methods for containment	Stop the flow of gas or remove cylinder to outdoor location if this can be done without risk. If leak is in container or container valve, contact the appropriate emergency telephone number in Section 1 or call your closest Indiana Oxygen location.
Methods for cleaning up	Return cylinder to Indiana Oxygen Company or an authorized distributor.
	7. HANDLING AND STORAGE
Precautions for safe handling	
Advice on safe handling	
	Protect cylinders from physical damage; do not drag, roll, slide or drop. Never attempt to lift a cylinder by its valve protection cap. When moving cylinders, even for short distance, use a cart designed to transport cylinders. Never insert an object (e.g. wrench, screwdriver, pry bar,etc.) into valve cap openings. Doing so may damage valve, causing leak to occur. Use an adjustable strap wrench to remove over-tight or rusted caps. If user experiences any difficulty operating cylinder valve discontinue use and contact supplier. Use only with adequate ventilation. Use only with equipment rated for cylinder pressure. Use a backflow preventive device in piping. Close valve after each use and when empty. Ensure the complete gas system has been checked for leaks before use.
	Never put cylinders into trunks of cars or unventilated areas of passenger vehicles. Never attempt to refill a compressed gas cylinder without the owner's written consent. Never strike an arc on a compressed gas cylinder or make a cylinder a part of an electrical circuit.
	Only experienced and properly instructed persons should handle gases under pressure. Always store and handle compressed gas cylinders in accordance with Compressed Gas Association, pamphlet CGA-P1, Safe Handling of Compressed Gases in Containers.
	For additional recommendations consult Compressed Gas Association's (CGA) Safety Bulletin SB-2, Oxygen-Deficient Atmospheres.
Conditions for safe storage, includ	ing any incompatibilities
Storage Conditions	Store in cool, dry, well-ventilated area of non-combustible construction away from heavily trafficked areas and emergency exits. Cylinders should be stored upright with valve protection cap in place and firmly secured to prevent falling. Keep at temperatures below 52°C / 125°F. Full and empty cylinders should be segregrated. Use a "first in-first out" inventory system to prevent full cylinders from being stored for excessive periods of time. Stored containers should be periodically checked for general condition and leakage.
Incompatible materials	Carbon dioxide is incompatible with:. Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.
8	. EXPOSURE CONTROLS/PERSONAL PROTECTION

# Control parameters

# Exposure Guidelines

Exposure Guidelines			
Chemical Name	ACGIH TLV	OSHA PEL	NIOSH IDLH
Carbon dioxide	STEL = 30000 ppm	TWA: 5000 ppm	IDLH: 40000 ppm
124-38-9	TWA: 5000 ppm	TWA: 9000 mg/m <sup>3</sup>	TWA: 5000 ppm
		(vacated) TWA: 10000 ppm	TWA: 9000 mg/m <sup>3</sup>
		(vacated) TWA: 18000 mg/m <sup>3</sup>	STEL: 30000 ppm
		(vacated) STEL: 30000 ppm	STEL: 54000 mg/m <sup>3</sup>

	(vacated) STEL: 54000 mg/m <sup>3</sup>	
	vernmental Industrial Hygienists - Threshold Limit Value. OSHA PEL: Occupational Safety and Health imits. NIOSH IDLH: Immediately Dangerous to Life or Health Immediately Dangerous to Life or Health.	
Other Information	Vacated limits revoked by the Court of Appeals decision in AFL-CIO v. OSHA, 965 F.2d 962 (11th Cir., 1992).	
Appropriate engineering controls		
Engineering Controls	Local exhaust ventilation to prevent accumulation of high concentrations and maintain air-oxygen levels at or above 19.5%. Oxygen detectors should be used when asphyxiating gases may be released. Systems under pressure should be regularly checked for leakages.	
Individual protection measures, such a	as personal protective equipment	
Eye/face protection	Wear safety glasses with side shields (or goggles).	
Skin and body protection	Work gloves and safety shoes are recommended when handling cylinders.	
Respiratory protection	Use positive pressure airline respirator with escape cylinder or self contained breathing apparatus for oxygen-deficient atmospheres (<19.5%). If exposure limits are exceeded or irritation is experienced, NIOSH/MSHA approved respiratory protection should be worn. Positive-pressure supplied air respirators may be required for high airborne contaminant concentrations. Respiratory protection must be provided in accordance with current local regulations.	
General Hygiene Considerations	Handle in accordance with good industrial hygiene and safety practice.	

# 9. PHYSICAL AND CHEMICAL PROPERTIES

Information on basic physical and chemical properties

Product Level Information:		
Physical state	Compressed gas	
Appearance	Colorless.	
Odor	Odorless.	
Odor threshold	No information available	
рН	No data available	
Melting point	No data available	
Evaporation rate	Not applicable	
Lower flammability limit:	Not applicable	
Upper flammability limit:	Not applicable	
Flash point	Not applicable.	
Autoignition temperature	No data available	
Decomposition temperature	No data available	
Partition coefficient	No data available	
Kinematic viscosity	Not applicable	

Chemical Name	Molecular weight	Boiling point	Vapor Pressure	Vapor density (air	Gas Density	Critical
				=1)	kg/m³@20°C	Temperature
Argon	39.95	-185.9 °C	Above critical temperature	1.38	1.65	-122.3 °C
Carbon dioxide	44.01	-78.5 °C (Sublimes)	838 psig (5778 kPa) @ 21.1°C	1.522	1.839	31.1 °C

# **10. STABILITY AND REACTIVITY**

Reactivity\_

Not reactive under normal conditions

# <u>Chemical stability</u>

Stable under normal conditions.

# Explosion data

Sensitivity to Mechanical Impact	None.
Sensitivity to Static Discharge	None.

Possibility of Hazardous Reactions None under normal processing.

<u>Conditions to avoid</u> Due to the presence of Carbon dioxide, Carbonic acid is formed in the presence of moisture.

# Incompatible materials

STOT - repeated exposure

Carbon dioxide is incompatible with:. Certain reactive metals, hydrides, moist cesium monoxide, or lithium acetylene carbide diammino may ignite. Passing carbon dioxide over a mixture of sodium peroxide and aluminum or magnesium may explode.

<u>Hazardous Decomposition Products</u> None known.

# **11. TOXICOLOGICAL INFORMATION**

# Information on likely routes of exposure

Inhalation	Acidosis, adrenal cortical exhaustion, and other metabolic stresses have resulted from prolonged continuous exposure to 1-2% carbon dioxide (10,000 ppm-20,000 ppm). The ACGIH TLV of 5,000 ppm is expected to provide a good margin of safety from asphyxiation and undue metabolic stress provided sufficient oxygen levels are maintained in the air. Increased physical activity, duration of exposure, and decreased oxygen content can affect systemic and respiratory effects resulting from exposure to carbon dioxide. Product is a simple asphyxiant.
Skin contact	No data available.
Eye contact	No data available.
Ingestion	Not an expected route of exposure.
Information on toxicological effects	
Symptoms	Simple asphyxiant. May cause suffocation by displacing the oxygen in the air. Exposure to oxygen-deficient atmosphere (<=18%) may cause dizziness, drowsiness, nausea, vomiting, excess salivation, diminished mental alertness, loss of consciousness and death. Exposure to atmospheres containing 8-10% or less oxygen will bring about unconsciousness without warning and so quickly that the individuals cannot help or protect themselves. Lack of sufficient oxygen may cause serious injury or death. Depending on concentration and duration of exposure to carbon dioxide may cause increased respirations, headache, mild narcotic effects, increased blood pressure and pulse, and asphyxiation. Symptoms of overexposure become more apparent when atmospheric oxygen is decreased to 15-17%.
Delayed and immediate effects as well	as chronic effects from short and long-term exposure
Irritation Sensitization Germ cell mutagenicity Carcinogenicity Reproductive toxicity Developmental Toxicity	Not classified. Not classified. Not classified. This product does not contain any carcinogens or potential carcinogens listed by OSHA, IARC or NTP. Not classified. Not classified.
STOT - single exposure	Not classified.

Not classified.

None known.

Central Vascular System (CVS), Respiratory system.

Chronic toxicity Target Organ Effects

Aspiration hazard		Not applicable.		
Numerical measures of toxici	ity			
Chemical Name	Oral LD50	Dermal LD50	Inhalation LC50	Inhalation LC50 (CGA P-20)
Carbon dioxide 124-38-9	-	-	470,000 ppm (Rat)	-
Product Information Oral LD50 Dermal LD50 Inhalation LC50	No informatic No informatic No informatic	on available		
	12. EC	COLOGICAL INFORI	MATION	
Ecotoxicity				
No known acute aquatic toxic				
Persistence and degradabilit Not applicable.	Y			
Bioaccumulation No information available.				
Global warming potential (G	WP) 1 (Carbon Dic	oxide)		
	13. DI	SPOSAL CONSIDER	ATIONS	
Waste treatment methods				
Disposal of wastes	container PR	OPERLY LABELED WITH AN	I waste or unused quantitie: Y VALVE OUTLET PLUGS OR C xygen Company for proper dis	APS SECURED AND VALVE
	14. TI	RANSPORT INFORM	MATION	
DOT				
UN/ID no.	UN1956			
Proper shipping name Hazard Class	Proper shipping name Compressed gas, n.o.s. Hazard Class 2.2			
Description	Description UN1956, Compressed gas, n.o.s. (Argon, Carbon Dioxide), 2.2			
Emergency Response Gu	ide Number 126			
<u>TDG</u>				
UN/ID no. Proper shipping name	uN1956 name Compressed gas, n.o.s.			
Hazard Class	2.2			
Description	UN1956, Com	UN1956, Compressed gas, n.o.s.(Argon, Carbon Dioxide), 2.2		
MEX				
LIN/ID no				

<u>^</u>	
UN/ID no.	UN1956
Proper shipping name	Compressed gas, n.o.s.
Hazard Class	2.2
Description	UN1956, Compressed gas, n.o.s. (Argon, Carbon dioxide), 2.2

# IATA

UN/ID no.	UN1956	
Proper shipping name	Compressed gas, n.o.s.	
Hazard Class	2.2	
ERG Code	2L	
Description	UN1956, Compressed gas, n.o.s. (Argon, Carbon dioxide), 2.2	
IMDG_		
UN/ID no.	UN1956	
Proper shipping name	Compressed gas, n.o.s.	
Hazard Class	2.2	
EmS-No.	F-C, S-V	
Special Provisions	274	
Description	UN1956, Compressed gas, n.o.s. (Argon, Carbon dioxide), 2.2	
ADR		
UN/ID no.	UN1956	
Proper shipping name	Proper shipping name Compressed gas, n.o.s.	
Hazard Class		
Classification code	ification code 1A	
Tunnel restriction code	inel restriction code (E)	
Special Provisions	274, 655	
Description	UN1956, Compressed gas, n.o.s. (Argon, Carbon dioxide), 2.2, (E)	

# **15. REGULATORY INFORMATION**

International Inventories	
TSCA	
DSL/NDSL	
EINECS/ELINCS	

#### Legend:

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List EINECS/ELINCS - European Inventory of Existing Chemical Substances/European List of Notified Chemical Substances

Complies Complies Complies

# **US Federal Regulations**

#### SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product does not contain any chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372.

SARA 311/312 Hazard Categories	
Acute Health Hazard	Yes
Chronic Health Hazard	No
Fire Hazard	No
Sudden release of pressure hazard	Yes
Reactive Hazard	No

# CERCLA

This material, as supplied, does not contain any substances regulated as hazardous substances under the Comprehensive Environmental Response Compensation and Liability Act (CERCLA) (40 CFR 302) or the Superfund Amendments and Reauthorization Act (SARA) (40 CFR 355). There may be specific reporting requirements at the local, regional, or state level pertaining to releases of this material.

# Clean Air Act, Section 112 Hazardous Air Pollutants (HAPs) (see 40 CFR 61)

This product does not contain any substances regulated as hazardous air pollutants (HAPS) under Section 112 of the Clean Air Act Amendments

# of 1990.

#### CWA (Clean Water Act)

This product does not contain any substances regulated as pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42)

#### **Risk and Process Safety Management Programs**

This material, as supplied, does not contain any regulated substances with specified thresholds under 40 CFR Part 68. This product does not contain any substances regulated as Highly Hazardous Chemicals pursuant to the 29 CFR Part 1910.110.

# US State Regulations

<u>California Proposition 65</u> This product does not contain any Proposition 65 chemicals

#### U.S. State Right-to-Know Regulations

Chemical Name	New Jersey	Massachusetts	Pennsylvania
Argon 7440-37-1	Х	X	Х
Carbon dioxide 124-38-9	Х	Х	Х

#### International Regulations

Chemical Name	Carcinogenicity	Exposure Limits
Carbon dioxide	-	Mexico: TWA= 5000 ppm
		Mexico: TWA= 9000 mg/m <sup>3</sup>
		Mexico: STEL= 15000 ppm
		Mexico: STEL= 27000 mg/m <sup>3</sup>

# **16. OTHER INFORMATION**

NFPA	Health hazards 0	Flammability 0	Instability 0	Physical and Chemical
		-	-	Properties Simple
				asphyxiant

Note: Ratings were assigned in accordance with Compressed Gas Association (CGA) guidelines as published in CGA Pamphlet P-19-2009, CGA Recommended Hazard Ratings for Compressed Gases, 3rd Edition.

Issue Date	07-Apr-2015
Revision Date	24-Jun-2015
Revision Note	Initial Release.

#### General Disclaimer

For terms and conditions, including limitation of liability, please refer to the purchase agreement in effect between Indiana Oxygen Company (or any of their affiliates and subsidiaries) and the purchaser.

# DISCLAIMER OF EXPRESSED AND IMPLIED WARRANTIES

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End of Safety Data Sheet